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Key indicators

Single-crystal X-ray study
 $T = 298\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.039
 wR factor = 0.093
Data-to-parameter ratio = 11.0For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

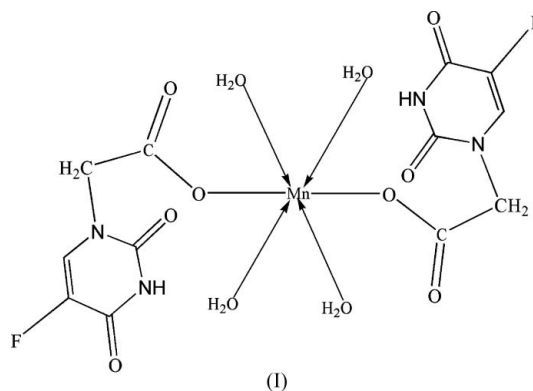
Tetraaquabis(5-fluorouracil-1-acetato)manganese(II)

In the centrosymmetric title compound, $[\text{Mn}(\text{C}_6\text{H}_4\text{N}_2\text{O}_4\text{F})_2(\text{H}_2\text{O})_4]$, each Mn^{II} ion is coordinated by two 5-fluorouracil-1-acetate anions *via* the carboxylate O atoms and four water molecules, forming a six-coordinate octahedral environment. $\text{N}-\text{H}\cdots\text{O}$ and $\text{O}-\text{H}\cdots\text{O}$ hydrogen-bonding interactions link adjacent molecules into a three-dimensional network.

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Comment

5-Fluorouracil (5-FU) is an antimetabolite with good antimicrobial and antitumor activity, but its administration is accompanied by significant toxic side effects and delivery problems (Ouchi *et al.*, 1997; Nichifor & Schacht, 1994; Nichifor *et al.*, 1997; Hulme *et al.*, 2005). In order to improve the topical delivery of 5-FU and reduce the side effects, many derivatives of 5-FU have been synthesized, some of which have better biological activity. 5-Fluorouracil-1-acetic acid (5-FUAA) is a member of this family (Sloan *et al.*, 1993; Li *et al.*, 2000; Beall & Sloan, 2001; Beall & Sloan, 2002). As increasing attention has been paid to the anticancer activity of 5-FU and its derivatives (Akgerman & Guney, 2000), several transition metal complexes have been reported (Wang *et al.*, 1993; Huang *et al.*, 2000; Hu *et al.*, 2005). The manganese derivative adduct, $\text{Mn}(\text{C}_6\text{H}_4\text{N}_2\text{O}_4\text{F})_2(\text{H}_2\text{O})_4$, (I), is reported here to build on these studies.



Mononuclear (I) consists of an Mn atom, four coordinated water molecules and two 5-fluorouracil-1-acetate anions, which bind through their carboxylate O atoms. The Mn atom lies on an inversion center and the geometry around the Mn ion is octahedral (Fig. 1 and Table 1). A square plane is formed by atoms O5, O5ⁱ, O6 and O6ⁱ [symmetry code: (i) $-x, -y, -z + 1$], which is crystallographically required to be planar. The $\text{N}-\text{H}\cdots\text{O}$ and $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds link the mononuclear units to form a three-dimensional network (Fig. 2 and Table 2).

Experimental

5-Fluorouracil-1-acetic acid (2 mmol, 0.75 g) and 1,3-di(4-pyridyl)propane (2 mmol, 0.40 g) were dissolved in a mixture of water (2 ml) and ethanol (8 ml). The solution was then added dropwise to a stirred aqueous solution (10 ml) of $\text{MnCl}_2 \cdot 2\text{H}_2\text{O}$ (1 mmol, 0.16 g). The resulting solution was filtered and allowed to evaporate slowly at room temperature. After four weeks, prismatic pink crystals of (I) appeared.

Crystal data

$[\text{Mn}(\text{C}_6\text{H}_4\text{N}_2\text{O}_4\text{F})_2(\text{H}_2\text{O})_4]$

$M_r = 501.23$

Monoclinic, $P2_1/c$

$a = 13.1820$ (13) Å

$b = 5.1106$ (5) Å

$c = 14.2202$ (14) Å

$\beta = 99.809$ (2)°

$V = 943.98$ (16) Å³

$Z = 2$

$D_x = 1.763$ Mg m⁻³

Mo $K\alpha$ radiation

Cell parameters from 2209

reflections

$\theta = 2.9$ – 25.2°

$\mu = 0.79$ mm⁻¹

$T = 298$ (2) K

Prism, pink

$0.32 \times 0.16 \times 0.13$ mm

Data collection

Bruker APEX area-detector diffractometer

φ and ω scans

Absorption correction: multi-scan (SADABS; Bruker, 2002)

$T_{\min} = 0.786$, $T_{\max} = 0.904$

4685 measured reflections

1687 independent reflections

1619 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$

$\theta_{\text{max}} = 25.2^\circ$

$h = -15 \rightarrow 15$

$k = -6 \rightarrow 5$

$l = -17 \rightarrow 12$

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.039$

$wR(F^2) = 0.093$

$S = 1.16$

1687 reflections

154 parameters

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0419P)^2 + 0.5989P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.19$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.26$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

Mn1—O5	2.1563 (18)	Mn1—O6	2.2148 (18)
Mn1—O1	2.1620 (16)		
O5—Mn1—O1 ⁱ	89.83 (7)	O5 ⁱ —Mn1—O6	92.57 (7)
O5—Mn1—O1	90.17 (7)	O1 ⁱ —Mn1—O6	85.81 (7)
O5—Mn1—O6	87.43 (7)	O1—Mn1—O6	94.19 (7)

Symmetry codes: (i) $-x, -y, -z + 1$.

Table 2

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{N2}-\text{H2} \cdots \text{O4}^{\text{ii}}$	0.86	1.99	2.843 (3)	175
$\text{O6}-\text{H6A} \cdots \text{O3}^{\text{iii}}$	0.83 (2)	2.02 (2)	2.834 (3)	170 (2)
$\text{O6}-\text{H6B} \cdots \text{O2}^{\text{i}}$	0.81 (2)	1.91 (2)	2.699 (3)	163 (2)
$\text{O5}-\text{H5A} \cdots \text{O1}^{\text{iv}}$	0.84 (2)	1.82 (2)	2.656 (2)	174 (2)
$\text{O5}-\text{H5B} \cdots \text{O3}^{\text{v}}$	0.82 (2)	1.98 (2)	2.788 (2)	170 (2)

Symmetry codes: (i) $-x, -y, -z + 1$; (ii) $-x + 1, -y, -z + 2$; (iii) $-x, y + \frac{1}{2}, -z + \frac{3}{2}$; (iv) $x, y - 1, z$; (v) $-x, y - \frac{1}{2}, -z + \frac{3}{2}$.

Water H atoms were located in difference density maps and refined with O—H and H \cdots H distances restrained to 0.82 (2) Å and 1.39 (1) Å, respectively, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{parent atom})$. The other H atoms were positioned geometrically and allowed to ride on their parent atoms at distances of $\text{Csp}^2-\text{H} = 0.93$ Å with $U_{\text{iso}}(\text{H}) =$

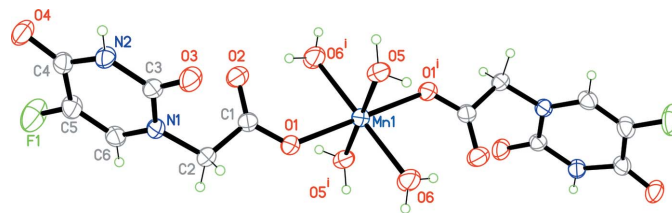


Figure 1

The molecular structure of (I), showing the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry code: (i) $-x, -y, -z + 1$.]

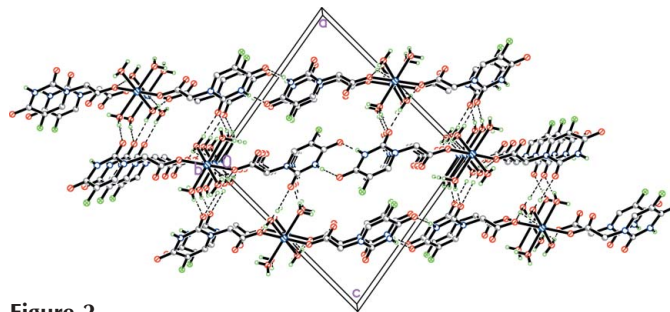


Figure 2

The three-dimensional network formed by hydrogen-bonding interactions, which are shown as dashed lines.

$1.2U_{\text{eq}}(\text{C})$, $\text{Csp}^3-\text{H} = 0.97$ Å with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$, and $\text{N}-\text{H} = 0.86$ Å with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2002); software used to prepare material for publication: SHELXL97.

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